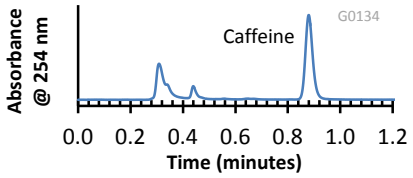
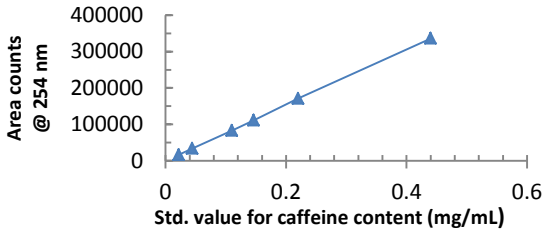
Application Note: 145-F

Determination of Caffeine in Soda Using HALO 5 C18

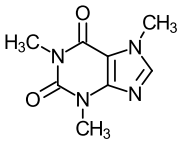




TEST CONDITIONS:

Column: HALO 5 C18, 3.0 x 50 mm, 5 µm with HALO 5 guard column
Part Number: HALO 5 C18, 95813-402
Part Number: Guard column: 95813-102
Mobile Phase: 75/25: A/B
A= 0.1% formic acid in water
B= Methanol
Flow Rate: 0.8 mL/min.
Pressure: 120 Bar
Temperature: 30°C
Injection Volume: 1.0 µL
Sample Solvent: (caffeine std.) mobile phase
Detection: UV 254 nm, VWD
Response Time: 0.02 sec.
Date rate: 25 Hz
Flow Cell: 2.5 µL semi-micro
LC System: Shimadzu Prominence UFLC XR
ECV: ~14 µL

STRUCTURE:



Caffeine

DISCUSSION:

A selection of sodas was purchased along with several energy drinks at a local grocery store. An approximate 10 mL sample from each soda was placed in a separate 20 mL scintillation vial and capped. The vials were shaken several times and then the cap was loosened and the vial sonicated for 10 minutes to remove CO_2. Then a 1 mL aliquot was placed into a 1.5 mL HPLC sample vial. A one microliter quantity was injected into the HPLC under conditions tabulated elsewhere on this page. A guard column was used to prevent the buildup of a brown material on the analytical column packing. The material was likely caramel coloring.

The chromatogram shown is from a regular cola drink.

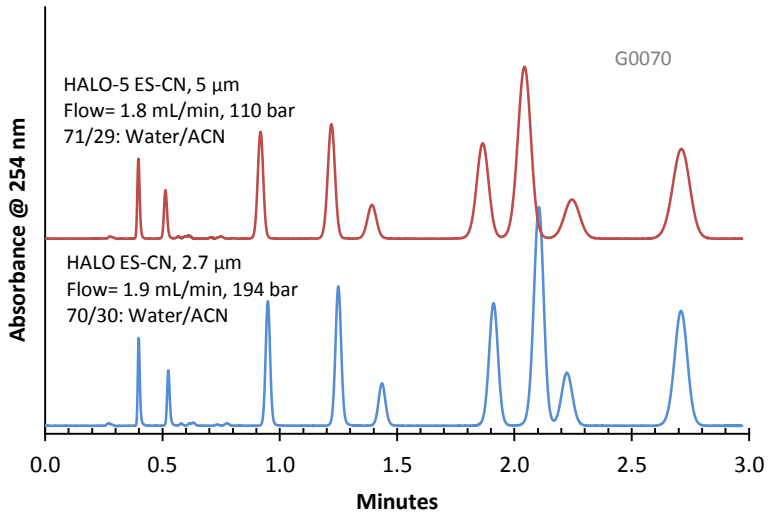
A standard curve of peak area vs. caffeine concentration was made over the range of 0.11-0.44 mg/mL. These values were used to calculate the line fit of Y=mX + b. From the peak response the concentration of (mg of caffeine)/mL was calculated and then multiplied by the number of mL in the beverage can (usually 355 mL). Results are shown below.

Sample	Caffeine tested mg/(355 mL)	Can value mg/(355 mL)
Store brand cola 1	12	N/A
Cola 2	53	54
Cola 3	43	43
Cola 4	36	38
Cola 5	38	38
Store brand diet cola 1	12	N/A
Diet cola 2	45	46
Diet cola 3	34	34
Diet cola 4	36	35
Energy drink 1*	160	160
Energy drink 2**	79	80
Diet Energy drink**	79	80
Non-cola drink 1	53.3	54
Non-cola drink 2	22	22
Diet non-cola drink	43	41
Diet cola 1 non caffeinated	0	N/A
Diet cola 2 non-caffeinated	0	N/A
Diet cola 3 non-caffeinated	0	N/A

355 mL = 12 oz.
*amount in 16 oz. (473 mL) cans
**amount in 8.4 oz (248 mL) cans

Application Note: 087-HA

Comparison of Selectivity of HALO ES-CN (2.7 µm) and HALO-5 ES-CN (5 µm) Phases



PEAK IDENTITIES:

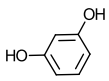
1. Resorcinol
2. Vanillin
3. Benzonitrile
4. Benzoin
5. Nitrobenzene
6. Benzanilide
7. Bisphenol A
8. Diethylphthalate
9. 3,4-Dinitrotoluene

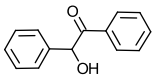
TEST CONDITIONS:

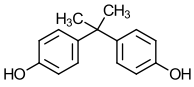
Column 1: 4.6 x 50 mm, HALO-5 ES-CN
Part Number: 95814-404
Column 2: 4.6 x 50 mm, HALO 2.7 ES-CN
Part Number: 92814-404
Mobile Phase: X/Y: Water/acetonitrile
See chart for X/Y ratio
Flow Rate: See chart
Pressure: See chart
Temperature: 30°C
Detection: UV 254 nm, VWD
Injection Volume: 1.0 µL
Sample Solvent: methanol
Response Time: 0.02 sec.
Flow Cell: 2.5 µL semi-micro
LC System: Shimadzu Prominence UFLC XR
ECV: ~14 µL

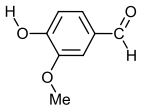
These chromatograms show the similarity in selectivity between the 5 µm and the 2.7 µm HALO ES-CN phases which allows the easy transfer of methods from one particle size packing to another.

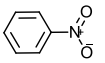
STRUCTURES:

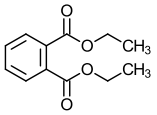

Resorcinol

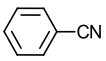

Benzoin

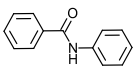

Bisphenol A

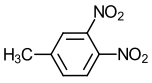

Vanillin


Nitrobenzene


Diethylphthalate


Benzonitrile


Benzanilide

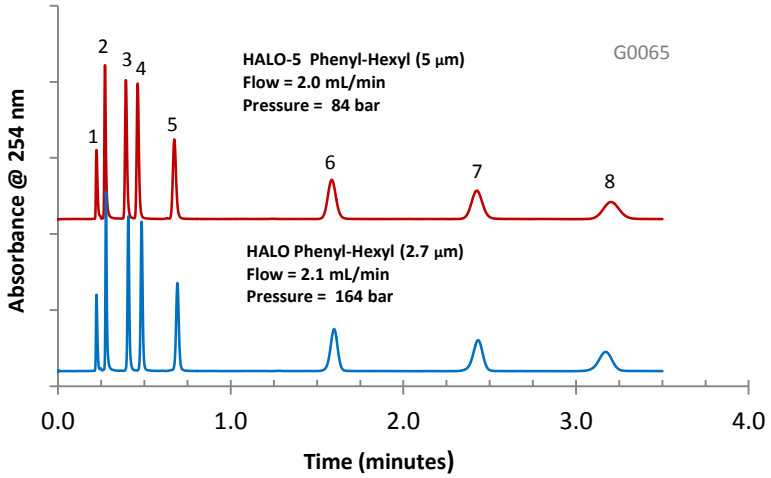

3,4-Dinitrotoluene

FOR MORE INFORMATION OR TO PLACE AN ORDER, CONTACT:

HALO® | Fused-Core® Particle Technology

Application Note: 082-HA

Comparable Selectivity between HALO-5 (5 µm) and HALO (2.7 µm) Phenyl-Hexyl Phases



PEAK IDENTITIES:

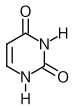
1. Uracil (t_0)
2. 6,7-Dihydroxycoumarin
3. 4-Hydroxycoumarin
4. Coumarin
5. 6-Chloro-4-hydroxycoumarin
6. Warfarin
7. Coumatetralyl
8. Coumachlor

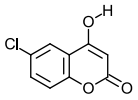
TEST CONDITIONS:

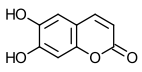
Column 1: 4.6 x 50 mm, HALO-5, 5 µm Phenyl-Hexyl
Part Number: 95814-406
Column 2: 4.6 x 50 mm, HALO 2.7 µm Phenyl-Hexyl
Part Number: 92814-406
Mobile Phase: A/B: 55/45
A= 0.1% Formic acid in water
B= 50/50: Methanol/acetonitrile
Flow Rate: See chart
Pressure: See chart
Temperature: 45°C
Detection: UV 254 nm, VWD
Injection Volume: 2 µL
Sample Solvent:
30/70: Water (0.1% formic acid)/methanol
Response Time: 0.12 sec.
Flow Cell: 5 µL
LC System: Agilent 1100

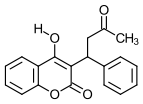
These chromatograms show the similarity in selectivity between the 5 µm and the 2.7 µm HALO Phenyl-Hexyl phases which allows the easy transfer of methods from one particle size to another.

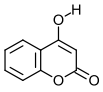
STRUCTURES:

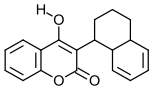

Uracil

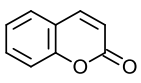

6-Chloro-4-hydroxycoumarin

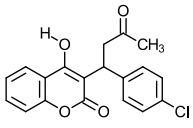

6,7-Dihydroxycoumarin


Warfarin


4-Hydroxycoumarin


Coumatetralyl


Coumarin


Coumachlor

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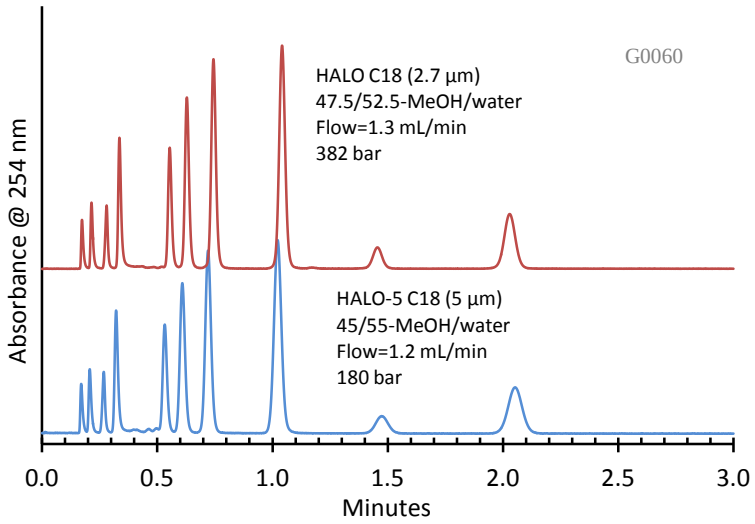
FOR MORE INFORMATION OR TO PLACE AN ORDER, CONTACT:

HALO® | Fused-Core® Particle Technology

Application Note: 077-HA

Comparable Selectivity of HALO C18 and HALO-5 C18



PEAK IDENTITIES:

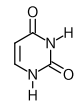
1. Uracil
2. Resorcinol
3. Aniline
4. 4-Chloroaniline
5. Acetoacetanilide
6. Dimethylphthalate
7. Cinnamyl alcohol
8. 2,6-Dinitrotoluene
9. Tolbutamide
10. 4-Chloro-3-nitroanisole

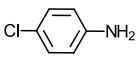
TEST CONDITIONS:

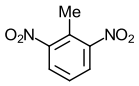
Column: 3.0 x 50 mm, HALO
Part Number: 92813-402, HALO C18 (2.7 µm)
Part Number: 95813-402, HALO-5 C18 (5 µm)
Mobile Phase: See figure.
Flow Rate: See figure.
Pressure: See figure
Temperature: 30°C
Detection: UV 254 nm, VWD
Injection Volume: 1.0 µL
Sample Solvent: Methanol
Response Time: 0.02 sec.
Flow Cell: 2.5 µL semi-micro
LC System: Shimadzu Prominence UFLC XR
ECV: ~14 µL

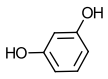
This mixture of compounds with varying functional groups and polarity show the same selectivity on both the 5-micron and 2.7-micron HALO C18 columns with only minor adjustments in flow rate and mobile phase composition being required. This separation demonstrates the ability to change from one HALO particle size to the other without needing to redevelop the method.

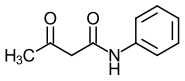
STRUCTURES:

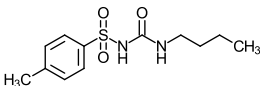

Uracil

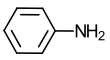

4-Chloroaniline

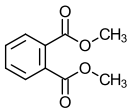

2,6-Dinitrotoluene

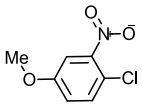

Resorcinol

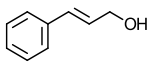

Acetoacetanilide


Tolbutamide


Aniline


Dimethylphthalate


4-Chloro-3-nitroanisole


Cinnamyl alcohol

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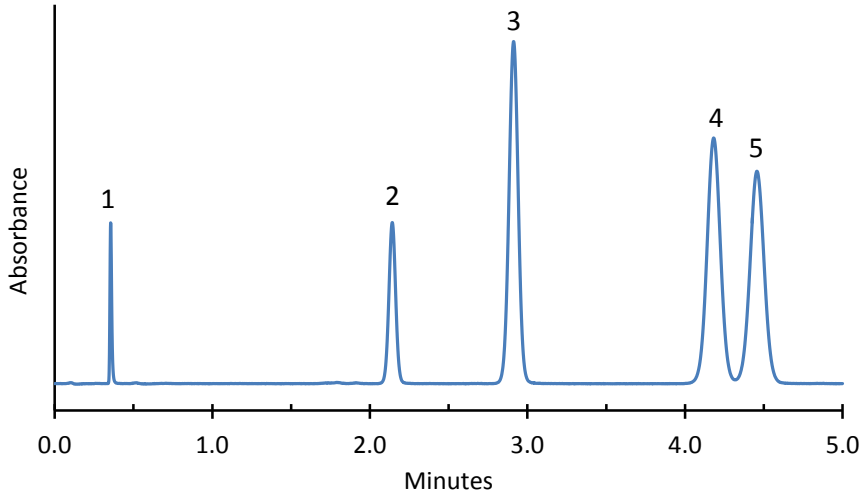
Application Note: 36-EX

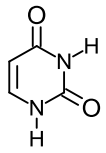
Isocratic Separation of Dinitrotoluenes on HALO PFP Phase



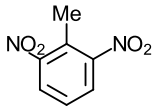
PEAK IDENTITIES:
1. Uracil
2. 2,6-Dinitrotoluene
3. 2,4-Dinitrotoluene
4. 3,4-Dinitrotoluene
5. 2,3-Dinitrotoluene

TEST CONDITIONS:
Column: 4.6 x 50 mm, HALO PFP
Part Number: 92814-409
Mobile Phase: 45/55-Water/Methanol
Flow Rate: 1.5 mL/min.
Pressure: 225 Bar
Temperature: 30 °C
Detection: UV 254 nm, VWD
Injection Volume: 1.0 µL
Sample Solvent: 50/50-Acetonitrile/Methanol
Response Time: 0.02 sec.
Flow Cell: 2.5 µL semi-micro
LC System: Shimadzu Prominence UFLC XR
Extra column volume: ~14 µL

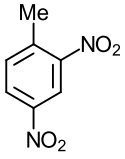
STRUCTURES:



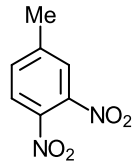
Uracil



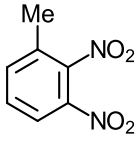
2,6-Dinitrotoluene



2,4-Dinitrotoluene



3,4-Dinitrotoluene

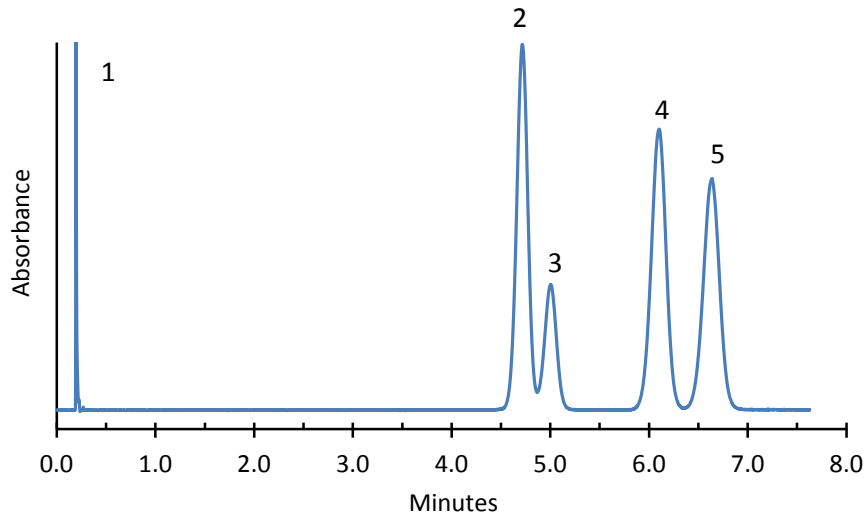


2,3-Dinitrotoluene

These dinitrotoluenes are difficult to separate, but can be separated with baseline resolution in under 5 minutes using a HALO Fused Core PFP (perfluorophenylpropyl) column.

Application Note: 35-EX

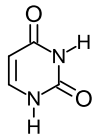
Isocratic Separation of Dinitrotoluenes on HALO RP-Amide Phase



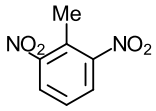
PEAK IDENTITIES:
1. Uracil
2. 2,4-Dinitrotoluene
3. 2,6-Dinitrotoluene
4. 3,4-Dinitrotoluene
5. 2,3-Dinitrotoluene

TEST CONDITIONS:
Column: 4.6 x 50 mm, HALO RP-Amide
Part Number: 92814-407
Mobile Phase: 80/20-Water/Acetonitrile
Flow Rate: 2.5 mL/min.
Pressure: 257 Bar
Temperature: 27 °C
Detection: UV 254 nm, VWD
Injection Volume: 1.0 µL
Sample Solvent: 50/50-Acetonitrile/Methanol
Response Time: 0.02 sec.
Flow Cell: 2.5 µL semi-micro
LC System: Shimadzu Prominence UFLC XR
Extra column volume: ~14 µL

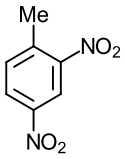
STRUCTURES:



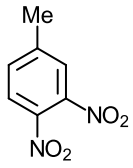
Uracil



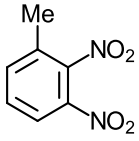
2,6-Dinitrotoluene



2,4-Dinitrotoluene



3,4-Dinitrotoluene



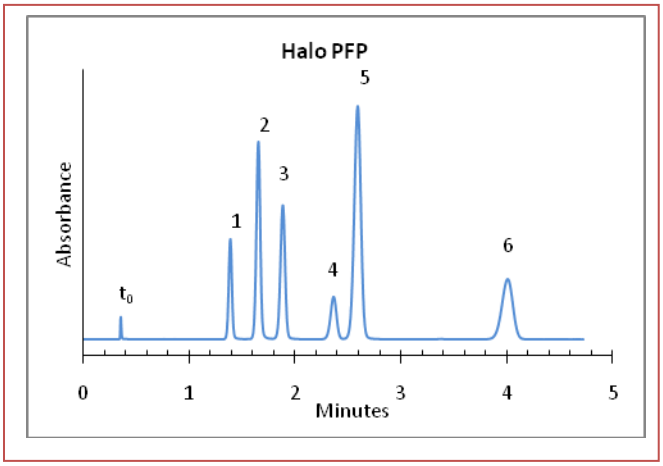
2,3-Dinitrotoluene

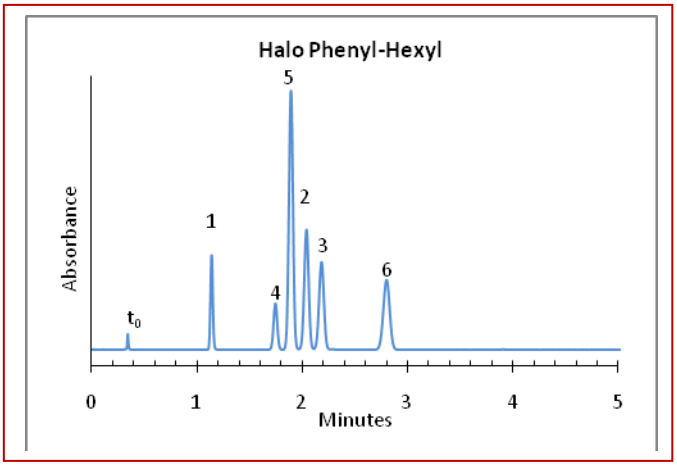
These dinitrotoluenes are difficult to separate, but can be separated with almost baseline resolution in under 7 minutes using a 50 mm long HALO Fused Core RP-Amide column.

HALO | Fused-Core® Particle Technology

Application Note: 26-P

Separation of Aromatic Nitro compounds on HALO PFP and Phenyl-Hexyl



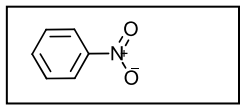


TEST CONDITIONS:
Column: 4.6 x 50 mm, HALO PFP, Phenyl-Hexyl
Part Numbers: 92814-409,-406, resp.
Mobile Phase: 45/55-water/methanol
Flow Rate: 1.5 mL/min.
Pressure: approximately 200 Bar
Temperature: 40 °C
Detection: UV 254 nm, VWD
Injection Volume: 0.5 µL
Sample Solvent: ~20/80-water/methanol
Response Time: 0.02 sec.
Flow Cell: 2.5 µL semi-micro
LC System: Shimadzu Prominence UFLC XR
Extra column volume: ~14 µL

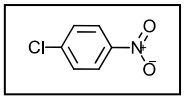
PEAK IDENTITIES:
1. Nitrobenzene
2. 1-Cl-4-Nitrobenzene
3. 2,6-Dinitrotoluene
4. 4-Nitrotoluene
5. 3-Nitrotoluene
6. 4-Cl-3-Nitroanisole

Differences in the interaction of the phenyl rings on the bonded phases with the pi electron systems of the nitro aromatic compounds result in significantly different selectivities that can be used to optimize these separations.

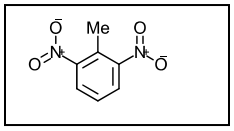
STRUCTURES:



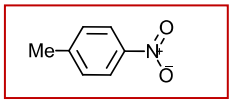
Nitrobenzene



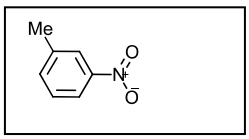
1-Chloro-4-Nitrobenzene



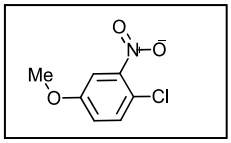
2, 6-Dinitrotoluene



4-Nitrotoluene



3-Nitrotoluene



4-Chloro-3-Nitroanisole

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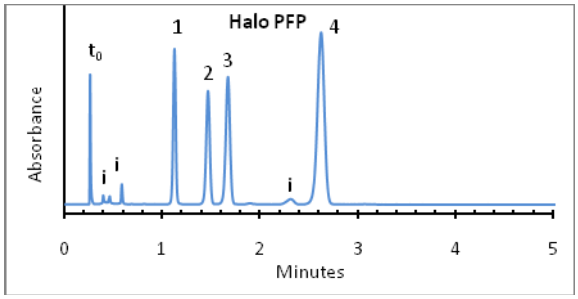
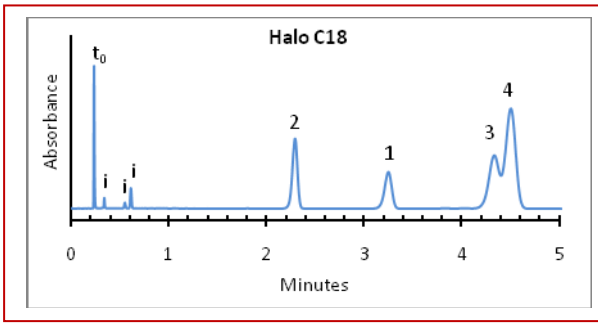
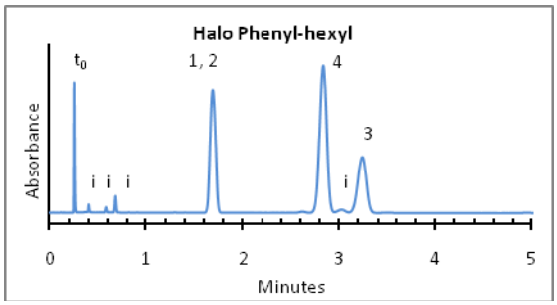
FOR MORE INFORMATION OR TO PLACE AN ORDER, CONTACT:

Application Note: 23-N

Separation of Neutral Aromatics on HALO PFP, C18 and Phenyl-Hexyl



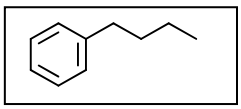




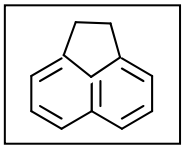
PEAK IDENTITIES:
1. Butylbenzene
2. Acenaphthene
3. 1-Phenylnaphthalene
4. Pyrene
5. I = impurities

TEST CONDITIONS:
Column: 4.6 x 50 mm, HALO PFP, C18, Phenyl-Hexyl
Part Numbers: 92814-409,-402,-406, respectively
Mobile Phase: 30/70-water/methanol
Flow Rate: 2.0 mL/min.
Pressure: approximately 250 Bar
Temperature: 40 °C
Detection: UV 254 nm, VWD
Injection Volume: 1.0 µL
Sample Solvent: methanol
Response Time: 0.02 sec.
Flow Cell: 2.5 µL semi-micro
LC System: Shimadzu Prominence UFLC XR
Extra column volume: ~14 µL

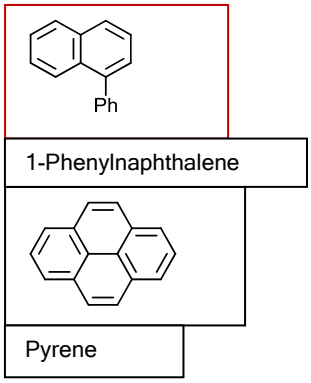
STRUCTURES:



Butylbenzene



Acenaphthene



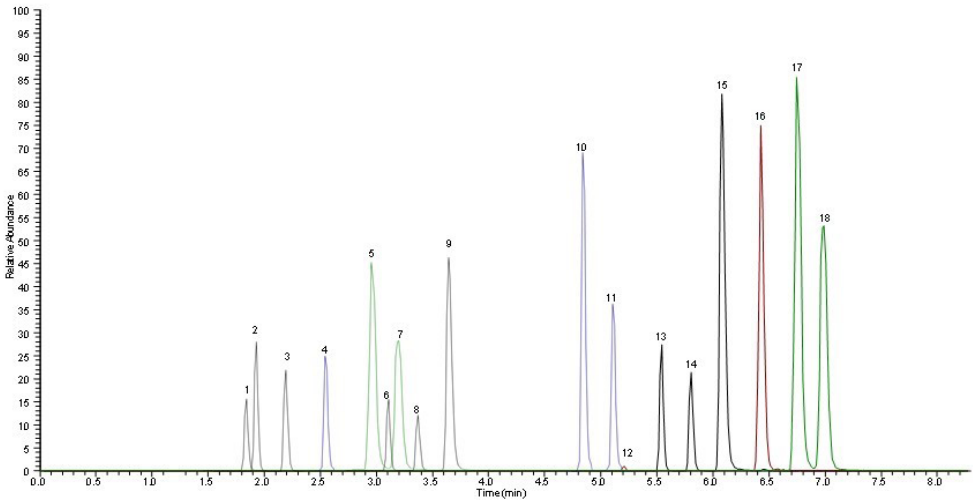
The separation of nonpolar aromatic compounds on these three Halo bonded phases under the same conditions show differences in selectivity that can be utilized in optimizing difficult separations.

HALO®

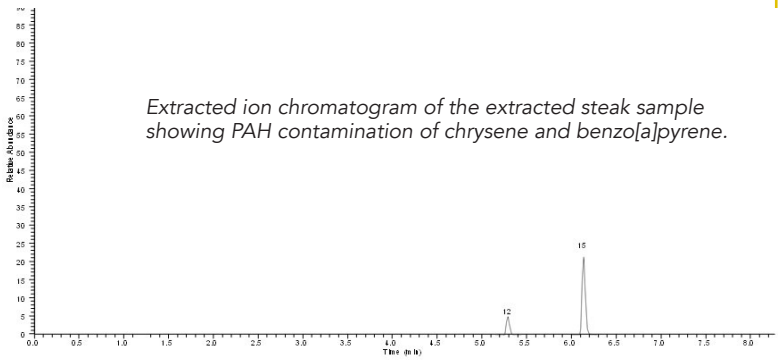
FOOD / BEVERAGE

LCMS of PAHs in Grilled Meat using HALO® PAH

237-P



TIC of a high resolution separation of 18 PAHs in under 7.5 minutes



Extracted ion chromatogram of the extracted steak sample showing PAH contamination of chrysene and benzo[a]pyrene.

PEAK IDENTITIES AND ELUTION ORDER

Peak #	Compound	Precursor Ion	Fragment 1	Fragment 2
1	Naphthalene	128	78	102
2	Acenaphthylene	152	126	151
3	1-Methylnaphthalene	142	89	115
4	2-Methylnaphthalene	142	115	141
5	Acenaphthene	154	126	153
6	Fluorene	166	115	165
7	Phenanthrene	178	151	176
8	Anthracene	178	152	176
9	Fluoranthene	202	150	200

Peak #	Compound	Precursor Ion	Fragment 1	Fragment 2
10	Pyrene	202	150	200
11	Benzo[a]anthracene	228	150	226
12	Chrysene	228	200	226
13	Benzo[b]fluoranthene	252	224	250
14	Benzo[k]fluoranthene	252	224	250
15	Benzo[a]pyrene	252	224	250
16	Dibenzo[a,h]anthracene	278	248	276
17	Benzo[ghi]perylene	276	248	274
18	Indeno[1,2,3-cd]pyrene	276	246	274

HALO®

FOOD / BEVERAGE

The HALO® PAH column continues in the tradition of HALO® products by offering high resolution separations, in high throughput time frames. 18 PAH compounds with 6 sets of isomeric compounds were able to be quickly and efficiently resolved in under 8 minutes. In addition, the high resolution separation of the HALO® PAH column, enabled chrysene and benzo[a]pyrene to be resolved from a complex meat matrix, enabling quantitation of PAH contamination present in barbequed steak. The concentration of PAHs in the sample, were below those established by the EU, and demonstrates that not only can the HALO® PAH column be used in the stringent regulatory testing of current established methods, but also be relied upon as future regulations dictate the establishment of new methods, requiring lower limits of detection. The HALO® PAH column offers a rugged and reproducible particle design meeting the needs of complex matrix testing. Fused-Core® technology is ideal for PAH analysis in particular, enabling customers to achieve analytical goals of speed, accuracy, and precision LC separations.

TEST CONDITIONS:

Column: HALO 90 Å PAH, 2.7 µm, 2.1 x 100 mm
Part Number: 92842-612
Flow Rate: 0.4 mL/min
Pressure: 289 bar
Column Temperature: 30 °C
Injection Volume: 1 µL
Sample Solvent: Methanol
LC System: Shimadzu Nexera
Mobile Phase **A:** Water/0.1% formic acid
B: Acetonitrile/0.1% formic acid

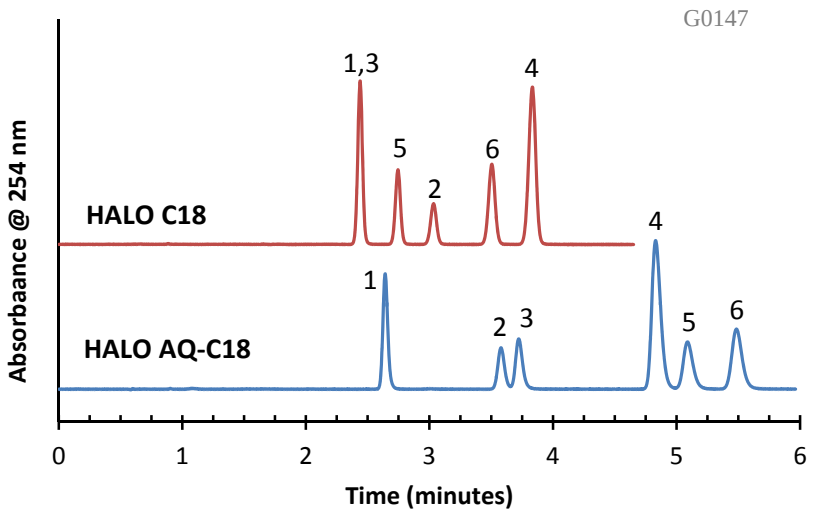
Gradient:	**Time**	**%B**
	0.0	40
	5.0	100
	8.0	100
	8.01	40

MASS SPECTROMETRY CONDITIONS:

MS System: Thermo Scientific™ Q Exactive™ HF
ESI voltage: 5.5 kV
Heater Temp: 400 °C
Sheath gas: 35 (arbitrary units)
Aux gas: 8 (arbitrary units)
Tube lens voltage: 40 V

Application Note: 157-G

Separation of Polar Samples on HALO AQ-C18 and C18



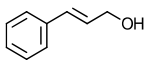
PEAK IDENTITIES:

1. Cinnamyl alcohol
2. 4'-Bromoacetanilide
3. Nitrobenzene
4. Anisole
5. 3,4-Dinitrotoluene
6. 2,4-Dinitrotoluene

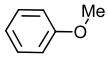
TEST CONDITIONS:

Columns: HALO C18, 4.6 x 100 mm, 2.7 µm
HALO AQ-C18, 4.6 x 100 mm, 2.7 µm
Part Numbers: 92814-602
92814-622
Mobile Phase: 48/52 (v:v): A/B
A = water
B = methanol
Flow Rate: 1.4 mL/min.
Pressure: HALO C18: 344 bar
HALO AQ-C18: 329 bar
Temperature: 30 °C
Detection: UV 254 nm, VWD
Injection Volume: 0.5 µL
Sample Solvent: methanol
Response Time: 0.02 sec.
Data rate: 25 Hz.
Flow Cell: 2.5 µL semi-micro
LC System: Shimadzu Prominence UFLC XR
ECV: ~14 µL

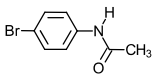
STRUCTURES:



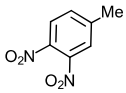
Cinnamyl alcohol



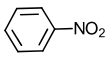
Anisole



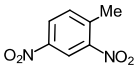
4'-Bromoacetanilide



3,4-Dinitrotoluene



Nitrobenzene



2,4-Dinitrotoluene

HALO AQ-C18 and HALO C18 phases have different selectivities as shown in the chromatograms above. The HALO AQ-C18 phase delivers increased retention for polar molecules compared to C18.

HALO®

TECHNICAL REPORT: AMT-TR062003

TITLE: **ANALYSIS OF POLYCYCLIC AROMATIC HYDROCARBONS IN GRILLED MEAT BY UHPLC/MS/MS**

MARKET SEGMENT: FOOD/BEVERAGE

AUTHOR:
Andrew Harron Ph.D., Application Scientist

ABSTRACT

A LCMS method was developed for the analysis of polycyclic aromatic hydrocarbons in meat samples following QuEChERS extraction. The HALO® PAH column demonstrated its performance and reliability by resolving 6 sets of isomeric PAHs in under 8 minutes. In addition, charbroiled meat samples (chicken and beef) were extracted and analyzed for PAH contamination, revealing quantitatible levels of PAHs in cooked steak.

INTRODUCTION

Polycyclic aromatic hydrocarbons (PAHs), formed from the incomplete combustion of fossil fuels, have been detected in foodstuffs throughout the world. In food, PAHs may be formed during industrial processing and domestic food preparation, such as barbecuing, smoking, drying, roasting, baking, frying or grilling. They can also enter food supply chains through contaminated air and water, and accumulate in various animals of food chains, such as livestock and fisheries. Listed as a confirmed human carcinogen by the US EPA, extensive exposure to PAHs by humans and animals has resulted in the development of cancer. PAHs, and various alkylated derivatives have a number of isomers and traditionally are analyzed by GC or GC-MS. However, because these isomers have to be chromatographically separated, the instrument run time is very long.

The analysis of PAHs by LCMS is challenging due to their low polarity and low solubility in aqueous environments, often requiring special post column derivation before introduction into the mass spectrometer. However, LCMS offers the ability for faster run times, provided that the isomeric species can be adequately separated. Fused-Core® technology is an ideal candidate to be used for PAH analysis, due to their high resolution and high throughput capabilities. Here we present an LCMS method using Fused-Core® technology to separate 18 PAHs in under 8 minutes and in addition the screening and quantitation of PAH compounds in cooked meat samples.

KEY WORDS:

PAHs, LCMS, Meat extraction, food safety, HALO® PAH column, superficially porous particles, Fused-Core® technology.

EXPERIMENTAL

PAH Standards, LCMS grade ACN, and QuEChERS kits were obtained from Millipore Sigma. Formic acid was obtained from Thermo Fisher. Charbroiled steak and chicken samples were prepared and extracted via the QuEChERS method. The column used was a HALO 90 Å PAH, 2.7 µm, 2.1 x 100 mm column (Advanced Materials Technology, Wilmington DE). Samples were run on a Shimadzu Nexera (Shimadzu Scientific Instruments, USA) coupled to an LTQ Velos Pro Orbitrap (Thermo Fisher Scientific, Bremen, Germany).

TEST CONDITIONS

Column: HALO 90 Å PAH, 2.7 µm, 2.1 x 100 mm
Part Number: 92842-612
Mobile Phase A: Water/0.1% formic acid
B: Acetonitrile/0.1% formic acid
Flow Rate: 0.4 mL/min
Gradient:

Time	%B
0.0	40
5.0	100
8.0	100
8.01	40

Pressure: 289 bar
Column Temperature: 30 °C
Injection Volume: 1 µL
Sample Solvent: Methanol
LC System: Shimadzu Nexera

MASS SPECTROMETRY CONDITIONS

Source Conditions	
ESI voltage	5.5 kV
Heater Temp	400 °C
Sheath gas (arbitrary units)	35
Aux gas (arbitrary units)	8
Tube lens voltage	40 V

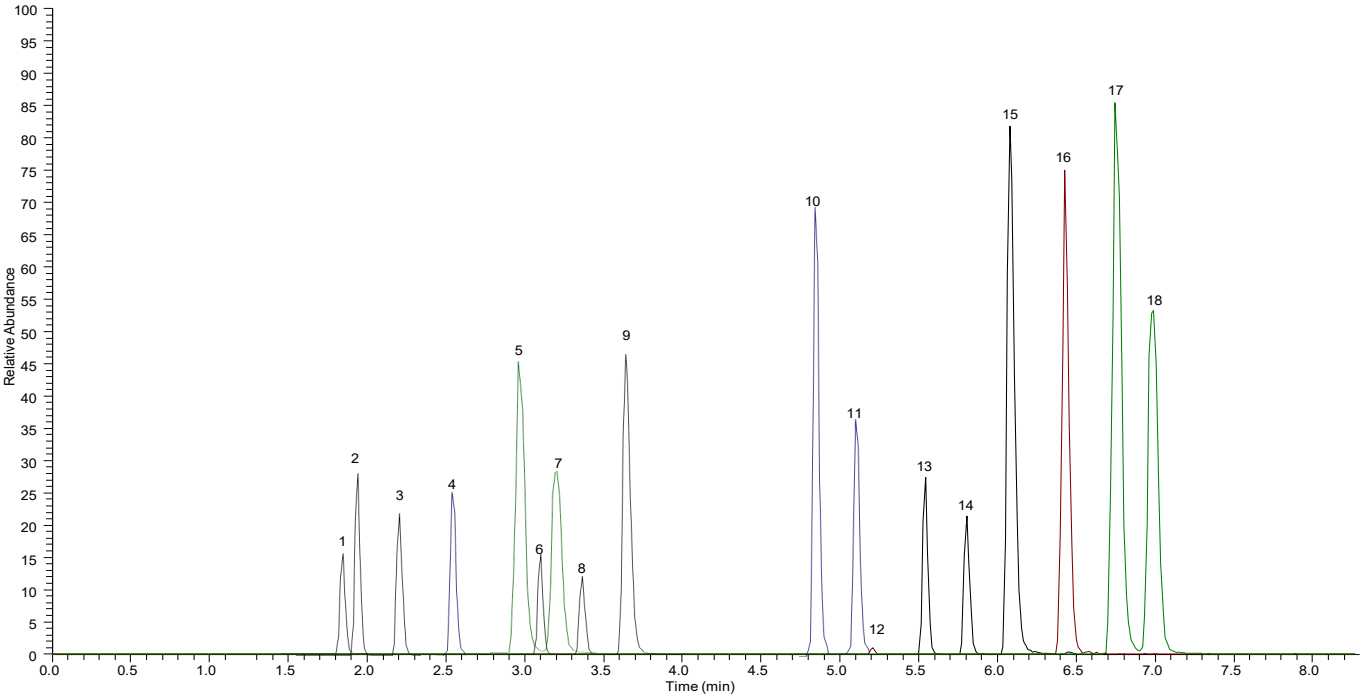


Figure 1. TIC of a high resolution separation of 18 PAHs in under 7.5 minutes

RESULTS

Using the HALO® PAH column 18 PAHs were able to be separated in under 7.5 minutes. Notice the clear separation of 6 sets of isomeric compounds (**Figure 1**). Table 1 shows the identities and elution order of the 18 compound mixture. Included in this table is a color code to show the isomeric compounds that were easily resolved by the HALO® PAH column, showing both high resolution and high throughput, which is a standard of HALO® performance.

Table 1. Peak identities and elution order

Compound	Precursor ion	Fragment 1	Fragment 2	Peak elution number
Naphthalene	128	78	102	1
Acenaphthylene	152	126	151	2
1-Methylnaphthalene	142	89	115	3
2-Methylnaphthalene	142	115	141	4
Acenaphthene	154	126	153	5
Fluorene	166	115	165	6
Phenanthrene	178	151	176	7
Anthracene	178	152	176	8
Fluoranthene	202	150	200	9
Pyrene	202	150	200	10
Benzo[a]anthracene	228	150	226	11
Chrysene	228	200	226	12
Benzo[b]fluoranthene	252	224	250	13
Benzo[k]fluoranthene	252	224	250	14
Benzo[a]pyrene	252	224	250	15
Dibenzo[a,h]anthracene	278	248	276	16
Benzo[ghi]perylene	276	248	274	17
Indeno[1,2,3-cd]pyrene	276	246	274	18

In order to test the performance and reliability of the HALO® PAH column, a screening and quantitation experiment was performed in which meat samples were investigated for PAH contamination following cooking. Charbroiled steak and chicken samples were obtained and then subjected to QuEChERS extraction. The extracted samples were then dried down, reconstituted in methanol and injected. No PAHs were detected in the chicken sample, in both the raw sample (data not shown) and in the cooked sample (**Figure 2**).

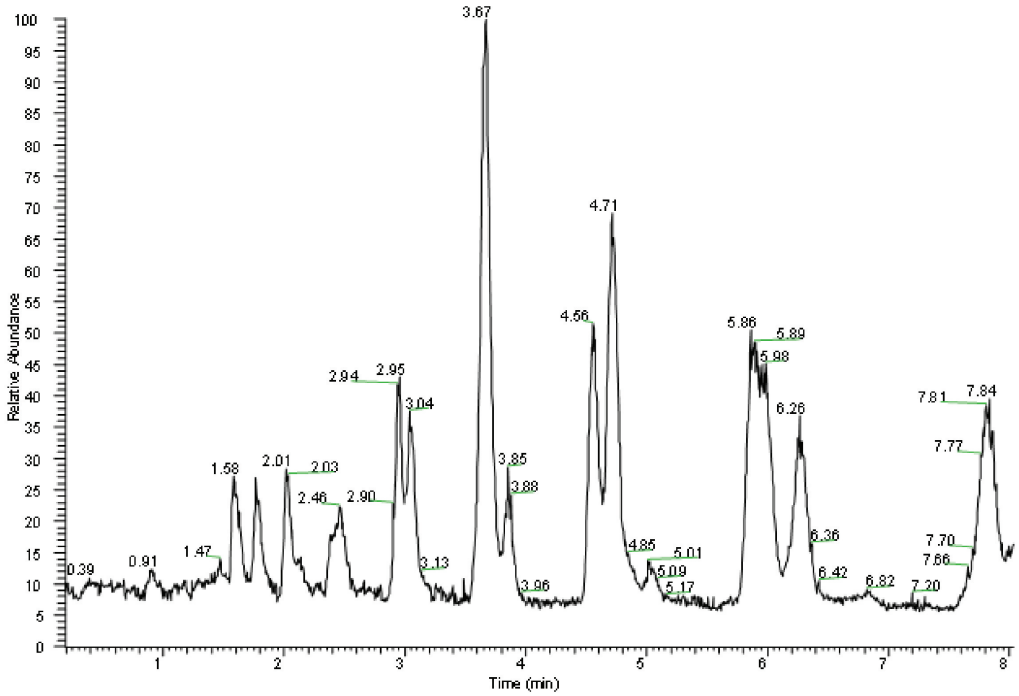


Figure 2. TIC of complex meat matrix of charbroiled chicken showing no PAHs detected

The steak sample, however, did indeed show evidence of PAH contamination (**Figure 3**), with chrysene and benzo[a]pyrene, found at detectable levels in the cooked steak sample.

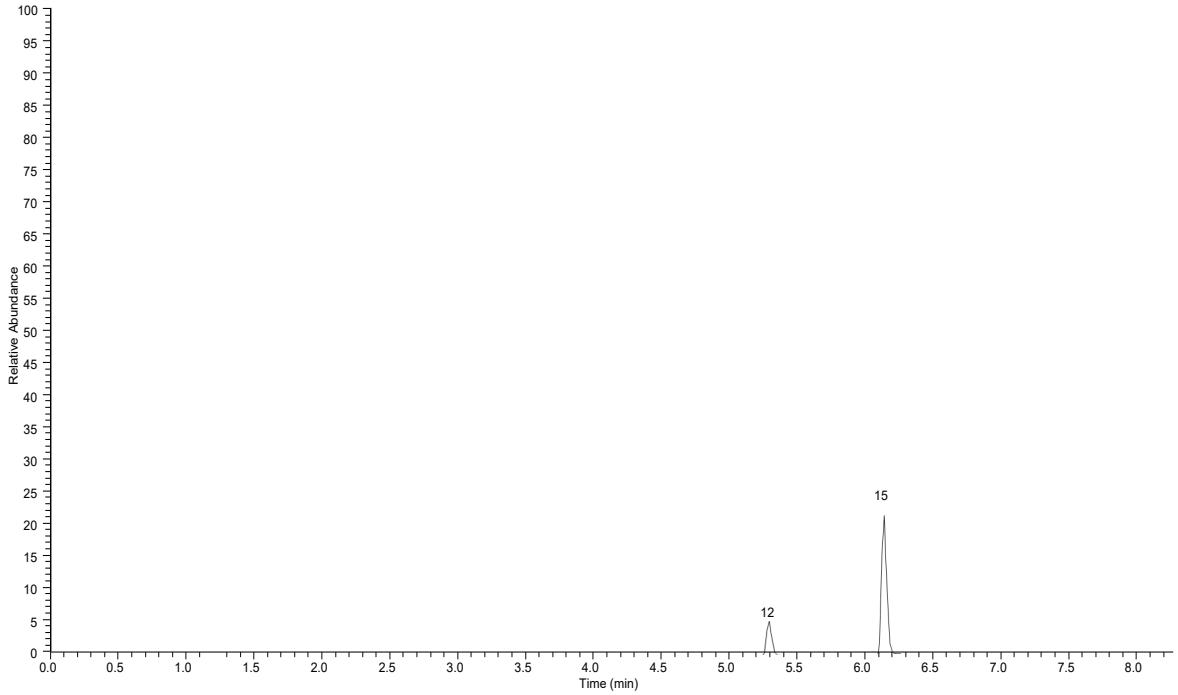


Figure 3. Extracted ion chromatogram of the extracted steak sample showing PAH contamination of chrysene and benzo[a]pyrene.

In order to determine the level of PAH contamination in the steak, a calibration curve was prepared at concentrations from 1 ppb-100 ppb, and the level of PAHs was quantified in the sample, with 1 ppb being the lower limit of detection.

Figure 4. Calibration curve of benzo [a] pyrene

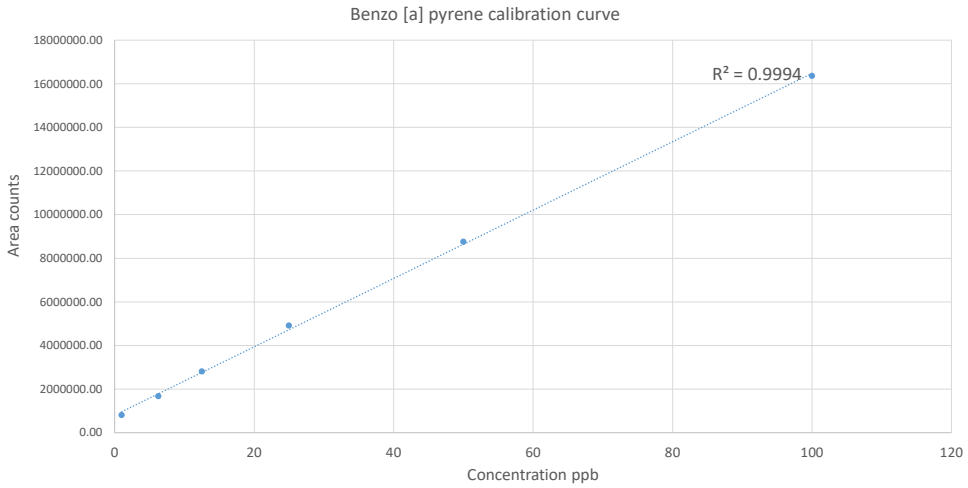
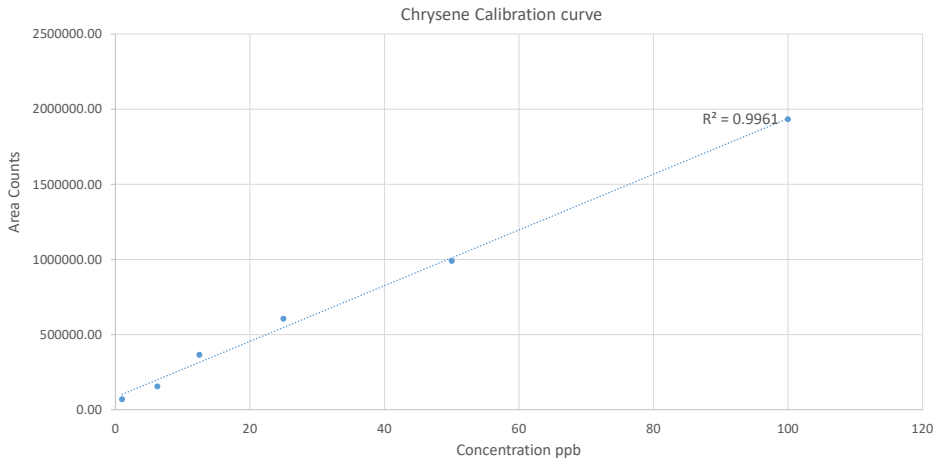


Figure 5. Calibration curve of chrysene



The concentration of benzo[a]pyrene was found to be 1.98 ppb and 2.55 ppb for chrysene (**Figures 4,5**). Smoke that is generated from fat that drips into the grill is the most likely explanation of the presence of these two PAHs. The EPA offers limited guidance on the maximum allowable limit of PAHs in food, however other entities such as the European Union have established limits. Maximum limits have been set by Commission Regulation (EC) No 1881/2006 for PAHs in key foodstuffs. The limit for cooked meat is 5 ppb for meat, and these results fall below those limits[1-3].

CONCLUSION

The HALO® PAH column continues in the tradition of HALO® products by offering high resolution separations, in high throughput time frames. 18 PAH compounds with 6 sets of isomeric compounds were able to be quickly and efficiently resolved in under 8 minutes. In addition, the high resolution separation of the HALO® PAH column, enabled chrysene and benzo [a] pyrene to be resolved from a complex meat matrix, enabling quantitation of PAH contamination present in barbequed steak. The concentration of PAHs in the sample, were below those established by the EU, and demonstrates that not only can the HALO® PAH column be used in the stringent regulatory testing of current established methods, but also be relied upon as future regulations dictate the establishment of new methods, requiring lower limits of detection. The HALO® PAH column offers a rugged and reproducible particle design meeting the needs of complex matrix testing. Fused-Core® technology is ideal for PAH analysis in particular, enabling customers to achieve analytical goals of speed, accuracy, and precision LC separations.

REFERENCES:

1. *Commission Regulation (EC) No 1881/2006 of 19 December 2006 setting maximum levels for certain contaminants in foodstuffs, as amended by Regulations 1126/2007 and 629/2008*
2. *Commission Regulation (EU) No 1327/2014 of 12 December 2014 amending Regulation (EC) No 1881/2006 as regards maximum levels of polycyclic aromatic hydrocarbons (PAHs) in traditionally smoked meat and meat products and traditionally smoked fish and fishery products*
3. *Regulation (EC) No 2065/2003 of the European Parliament and of the Council of 10 November 2003 on smoke flavourings used or intended for use in or on foods*