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# TECHNICAL NOTE 2. Preparation of mobile phase for HPLC

## 1) Organic solvent/ aqueous mixed mobile phase

## 1)-1. Preparation of methanol : water = 70 : 30 (v/v) 1L

1 Measure 700 ml of methanol in a measuring cylinder.

O Measure 300 ml of distilled water in a measuring cylinder.

3 Mix 1 and 2 thoroughly and degas.

The better approach is to prepare the mobile phase gravimetrically rather than volumetrically. Following is example of preparation.

Composition	table for	mobile phas	e 1L	using	methanol	and	water
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Methanol / Water	Methanol (g)	Distilled water (g)	
90/10 (v/v)	711.9	99.8	
80/20 (v/v)	632.8	199.6	
70/30 (v/v)	553.7	299.5	
60/40 (v/v)	474.6	399.3	
50/50 (v/v)	395.5	499.1	
40/60 (v/v)	316.4	598.9	
30/70 (v/v)	237.3	698.7	
20/80 (v/v)	158.2	798.6	
10/90 (v/v)	79.1	898.4	

Composition table for mobile phase 1L using acetonitrile and water

Acetonitrile / Water	Acetonitrile (g)	Distilled water (g)
90/10 (v/v)	707.4	99.8
80/20 (v/v)	628.8	199.6
70/30 (v/v)	550.2	299.5
60/40 (v/v)	471.6	399.3
50/50 (v/v)	393.0	499.1
40/60 (v/v)	314.4	598.9
30/70 (v/v)	235.8	698.7
20/80 (v/v)	157.2	798.6
10/90 (v/v)	78.6	898.4

Caution : Methanol and acetonitrile are hazardous substances, do not use for medical purpose. Always process in a laboratory hood and wear an eye protection and a mask.





 Column
 COSMOSIL 5C18-MS-II

 Column size
 4.6 mm I.D. × 150 mm

 Flow rate
 1.0 ml/min

 Detection
 254 nm 0.16 AUFS

 Temperature
 30°C

 Sample
 1 uracil

 2. naphthalene

Special attention should be paid to measure correct amount of organic solvent as the retention time is significantly changed by 1% different composition.

Differences between acetonitrile and methanol in reversed phase liquid chromatography





## 2) Organic solvent/ buffer mixed mobile phase

- 2)-1. Preparation of methanol : 20 mmol/l phosphate buffer (pH2.5) = 80 : 20 (v/v) 1L
  - Preparation of 20 mmol/l phosphate buffer (pH2.5)
  - 1 Prepare 20 mmol/l sodium dihydrogenphosphate aqueous solution.
  - ② Prepare 20 mmol/l phosphoric acid aqueous solution.
  - 3 Adjust the pH to 2.5 by mixing 1 with 2.
  - ④ Filter ③ under reduced pressure to remove insoluble substance that may deteriorate pump-seal and clog columns (0.45 μm or smaller pore size is recommended).

### (Easy method)

- ① Dissolve 1.31 g of sodium dihydrogenphosphate and 1.05 g of phosphoric acid in distilled water to make 1 L solution.
- (2) Filter the solution under reduced pressure to remove insoluble substance (0.45  $\mu$ m or smaller pore size is recommended).
- 3 Confirm that the solution is pH2.5.
- Preparation of methanol : 20 mmol/l phosphate buffer (pH2.5) = 80 : 20 1L
- ① Measure 800 ml of methanol in a measuring cylinder.
- 0 Measure 200 ml of 20 mmol/l phosphate buffer (pH2.5) in a measuring cylinder.
- 3 Mix 1 and 2 thoroughly and degas.

### 2)-2. Preparation of methanol : 20 mmol/l phosphate buffer (pH7.0) = 80 : 20 (v/v) 1L

- Preparation of 20 mmol/I phosphate buffer (pH7.0)
- ① Prepare 20 mmol/l sodium dihydrogenphosphate aqueous solution.
- (2) Prepare 20 mmol/l di-sodium hydrogenphosphate aqueous solution.
- 3 Adjust the pH7.0 by mixing 1 with 2 .
- ④ Filter ③ under reduced pressure to remove insoluble substance that may deteriorate pump-seal and clog columns (0.45 μm or smaller pore size is recommended).

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(Easy method)

- ① Dissolve 1.14 g of sodium dihydrogenphosphate and 1.49 g of di-sodium hydrogenphosphate in distilled water to make 1L solution.
- 2 Filter the solution under reduced pressure to remove insoluble substance (0.45  $\mu$ m or smaller pore size is recommended).
- 3 Confirm that the solution is pH7.0.
- Preparation of Methanol : 20 mmol/l phosphate buffer (pH7.0) = 80 : 20 1 L
- 1 Measure 800 ml of methanol in a measuring cylinder.
- 2 Measure 200 ml of 20 mmol/l phosphate buffer (pH7.0) in a measuring cylinder.
- 3 Mix 1 and 2 thoroughly and degas.

The better approach is to prepare the mobile phase gravimetrically rather than volumetrically. Following is example of preparation.

methanol : 20 mmol/l phosphate buffer	methanol (g)	20 mmol/l phosphate buffer (pH2.5) (g)	20 mmol/l phosphate buffer (pH7.0) (g)
90/10 (v/v)	711.9	99.8	99.9
80/20 (v/v)	632.8	199.6	199.8
70/30 (v/v)	553.7	299.4	299.7
60/40 (v/v)	474.6	399.2	399.6
50/50 (v/v)	395.5	499.0	499.5
40/60 (v/v)	316.4	598.8	599.4
30/70 (v/v)	237.3	698.6	699.3
20/80 (v/v)	158.2	798.4	799.2
10/90 (v/v)	79.1	898.2	899.1

Caution : Methanol and acetonitrile are hazardous substances, do not use for medical purpose. Always process in a laboratory hood and wear an eye protection and a mask.

### 3) Preparation of ion pair reagent containing mobile phase

3)-1.Preparation of 5 mmol/l sodium 1-butanesulfonate containing 20 mmol/l phosphate buffer (pH2.5)

- ① Prepare 5 mmol/l sodium 1-butanesulfonate containing 20 mmol/l sodium dihydrogenphosphate aqueous solution.
- ② Prepare 5 mmol/I sodium 1-butanesulfonate containing 20 mmol/I phosphoric acid aqueous solution
- 3 Adjust the pH to 2.5 by mixing 1 with 2 .
- ④ Filter ③ under reduced pressure to remove insoluble substance that may deteriorate pump-seal and clog columns (0.45 μm or smaller pore size is recommended).

(Easy method)

- ① Dissolve 1.31 g of sodium dihydrogenphosphate, 1.05 g of phosphoric acid and 0.80 g of sodium 1-butanesulfonate in distilled water to make 1L solution.
- (2) Filter the solution under reduced pressure to remove insoluble substance (0.45  $\mu$ m or smaller pore size is recommended).
- 3 Confirm that the solution is pH2.5.

% 0.5M sodium 1-butanesulfonate aqueous solution is also available from Nacalai Tesque.