

# Titanium Dioxide – Properties and Applications in LC

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## Abstract

For decades science and industry have been searching for sorbents useful in liquid chromatography with the following properties:

- Excellent stability at pH values between pH 1 and pH 14
- Symmetric and narrow peaks for basic compounds along with
- High selectivity
- High mechanical stability and
- Low column back pressures

This sorbent is now commercially available under the trade name Sachtopore.

Chemically, Sachtopore is a porous, ceramic titanium dioxide in the anatase or rutile modification. Titanium dioxide has acid/base properties intermediate between silicas and aluminum oxides. Unlike these, however, it is practically insoluble in bases as well as acids. It exhibits very selective adsorption/desorption behaviour for organic molecules, especially for bases and isomers.

The poster presents characteristic examples for separations that are impossible to perform with inorganic stationary phases known so far. Furthermore, examples for other benefits – such as its durability in cleaning in place (CIP) procedures – are shown.

## Introduction

By choosing suitable stationary phases and separation conditions the performance and productivity of a large scale separation process can be improved significantly. Porous titanium dioxide – namely Sachtopore™ – is different in selectivity compared to silica and alumina which are the most commonly used oxide packings for NP-chromatography. This difference offers a high innovative potential in order to gain more selective separations and enhance productivity.

## General Features of Sachtopore

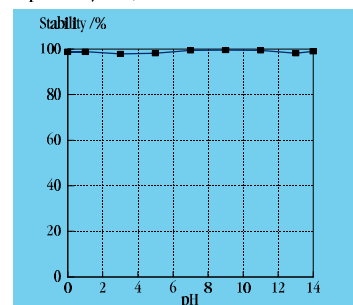
- Robust
- Spherical particle shape (Fig. 1)
- 100% crystalline
- Interconnected crystals

## Physical Properties of Sachtopore

Sachtopore	60 Å	100 Å	300 Å	2000 Å
Pore diameter [Å]	60	100	300	2000
Specific surface area [m <sup>2</sup> /g]	100	55	15	<5
Specific pore volume [ml/g]	0.2	0.21	0.12	n. d.
Particle size available	3, 5, 10, 20, 40, 80 µm, 1 – 2 mm			
<b>Mechanical stability</b>				
Applicable packing pressure [bar]	600	600	600	600
Applicable piston pressure in DAC™ columns [bar]	100	100	100	80

## Chemical Properties of Sachtopore

- Pharmaceutical grade purity
- Homogeneous surface chemistry
- pH-stability 1–14, stable to CIP in 1 M NaOH



- Moderate Lewis acidity
- Strong adsorption of multidentate ligands which bind specifically to Ti(IV)

## Chromatographic Properties of Sachtopore

- Scalable from analytical to preparative
- Basic and acidic modifier applicable in NP-mode (Fig. 2)

## Extraordinarily useful for:

- Separation of basic compounds (Fig. 3)
- Large scale separations of isomers (Fig. 5)
- Purification of basic pharmaceuticals (Fig. 5, 6)

## Conclusion

Sachtopore is an ideal packing material for NP-separations which are difficult or even impossible to perform on other stationary phases. The unique selectivity of Sachtopore can enhance the selectivity coefficient and can dramatically improve the productivity of the entire large scale separation process. Thus Sachtopore can be much more than an alternative to commonly used packings; it is a serious candidate to substitute them.

## References

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- A. Kurganov, U. Trüding, T. Isacova and K. Unger, Chromatographia, Vol. 42 No. 34 (1996) 217

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DAC is a trademark of Prochrom, France  
Sachtopore is a trademark of Sachleben, Germany

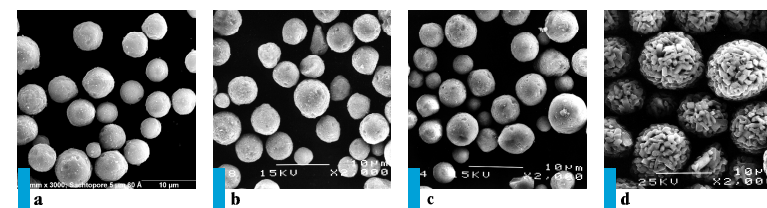


Fig. 1: SEMs of porous Titanium Dioxide beads (a: 60 Å, b: 100 Å, c: 300 Å, d: 2000 Å)

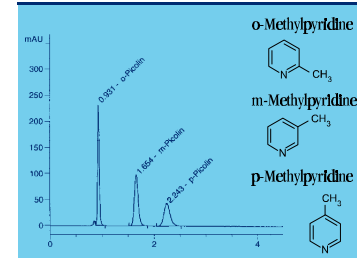


Fig. 2: Normal Phase separation of isomeric picolines using a basic modifier.<sup>c</sup>

Column: Porous Titanium Dioxide (100 Å, 5 µm, 150 x 4 mm)  
Eluent: n-Heptane/Ethanol (NH<sub>3</sub>)\* = 97/3 (v/v)  
\*Ethanol/NH<sub>3</sub> (25%) = 1/500  
Flow: 1.5 ml/min  
Detection: UV @ 210 nm  
Temperature: 43 °C  
Injection: 2 µl

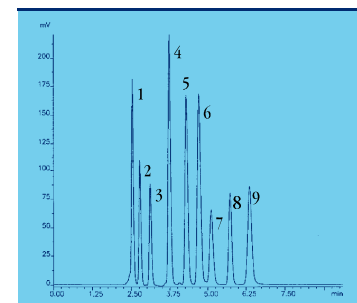
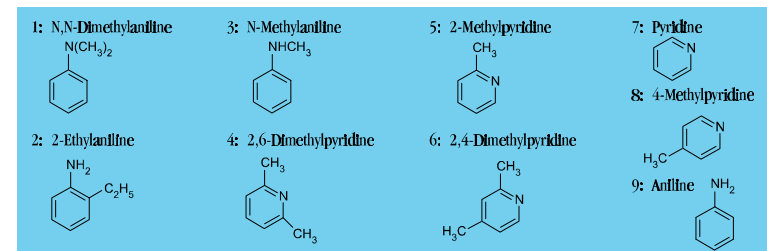


Fig. 3: Normal Phase separation of basic sample solutes.<sup>d</sup>

Column: Porous Titanium Dioxide (100 Å, 5 µm, 150 x 4 mm)  
Eluent: n-Heptane/2-Propanol = 99.5/0.5 (v/v)  
Flow: 0.75 ml/min  
Detection: UV @ 254 nm  
Temperature: ambient  
Pressure: 16 bar



<sup>c</sup> with courtesy of H. Schmidt, Prof. Dr. B. Wencławiak, Universität Gesamthochschule Siegen, Dr. E. Weber, Bayer AG

<sup>d</sup> with courtesy of Prof. Dr. A. Kurganov, Universität Mainz

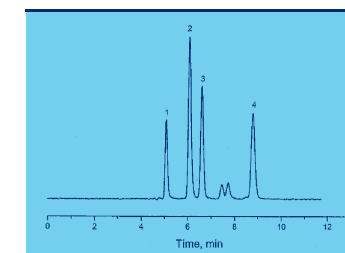


Fig. 4: Separation of unpolar sample solutes.<sup>d</sup>

Column: Porous Titanium Dioxide (100 Å, 5 µm, 250 x 4 mm)  
Eluent: n-Heptane  
Flow: 0.5 ml/min  
Detection: UV @ 230 nm  
Temperature: ambient  
Pressure: 33 bar

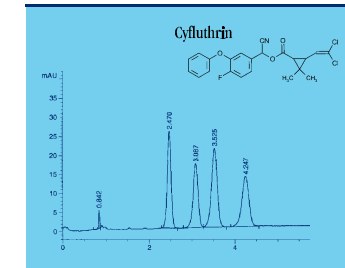
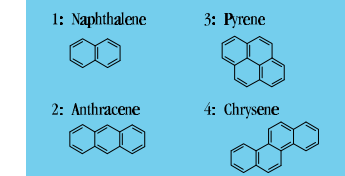


Fig. 5: Normal Phase separation of isomeric pyrene derivatives.<sup>d</sup>

Column: Porous Titanium Dioxide (100 Å, 5 µm, 150 x 4 mm)  
Eluent: n-Heptane/MTBE = 80/20 (v/v)  
Flow: 1.5 ml/min  
Detection: UV @ 220 nm  
Temperature: 43 °C  
Injection: 2 µl

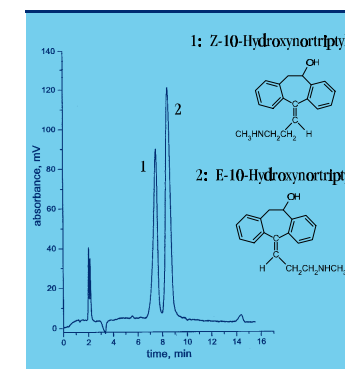


Fig. 6: Normal Phase separation of basic sample solutes.<sup>d</sup>

Column: Porous Titanium Dioxide (100 Å, 5 µm, 250 x 4 mm)  
Eluent: Dichloromethane/Methanol = 96.5/3.5 (v/v)  
Flow: 1.0 ml/min  
Detection: UV @ 254 nm  
Temperature: ambient