

A Practical, Selectivity Based Hydrophilic Interaction Liquid Chromatography (HILIC) Method Development Protocol

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1. Introduction

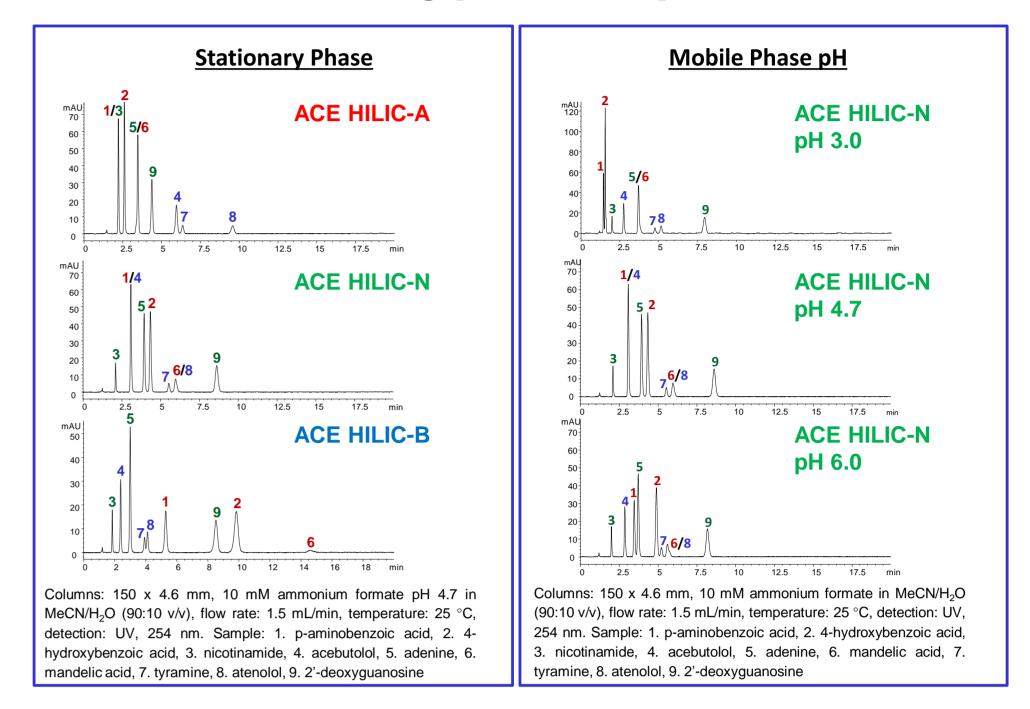
- Exploring chromatographic selectivity is a powerful approach for LC method development.
- Efficient method development requires a logical exploration of the key chromatographic parameters affecting selectivity.
- Rationally designed method development strategies
 assess key parameters and allow well informed
 decision making leading to robust stationary phase /
 mobile phase selection.

2. Selectivity in HILIC

- In HILIC, the column stationary phase has a significant effect on chromatographic selectivity.
- The ACE HILIC range consists of three complementary phases specifically designed to offer maximum selectivity differences – ideal for method development:

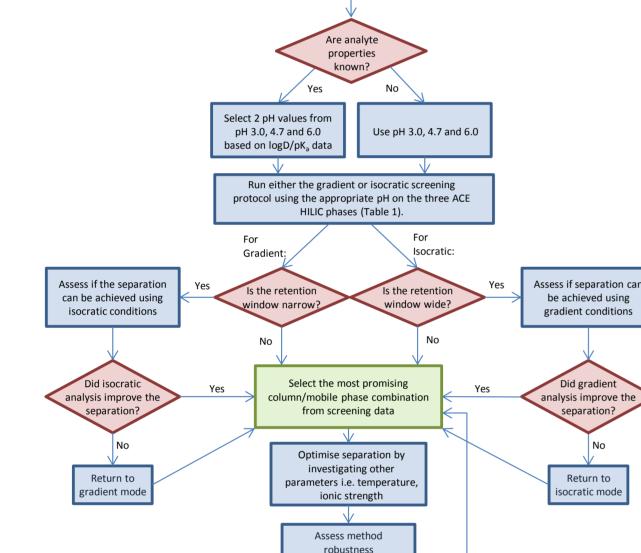
ACE HILIC-B

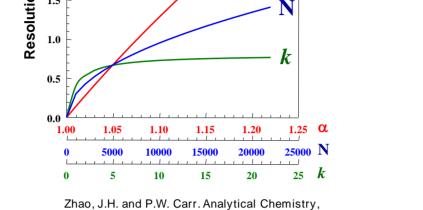
3. Effect of stationary phase and pH



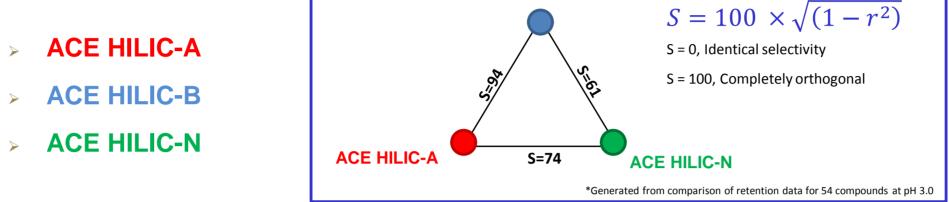
- Method development strategies based on column / mobile phase screening and optimisation are commonly utilised for reversed phase.
- For HILIC, such strategies are less common / less well defined.
- This poster demonstrates a simple, step-by-step approach to HILIC method development, based on the concept of exploring column selectivity.







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- Mobile phase pH is also a powerful parameter and can affect ionisation of analytes and the stationary phase itself.
- Method development strategies based on screening different stationary phases and mobile phase pH are therefore the optimum choice.
- Buffer concentration and temperature are less influential, however can be used to fine-tune methods.

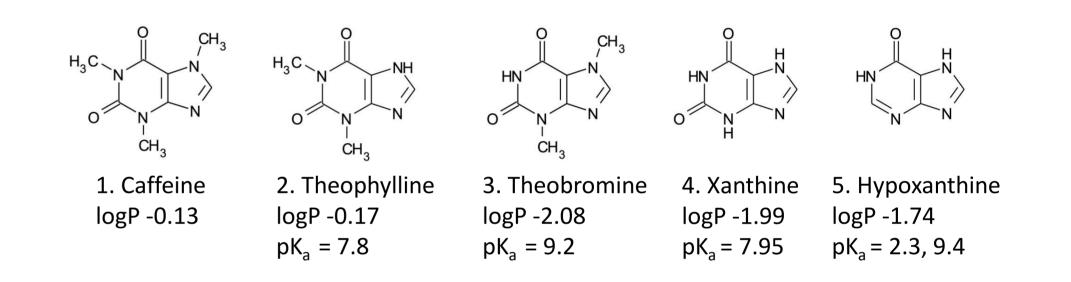
5. Screening Conditions

Isocratic and gradient screening is performed as follows:

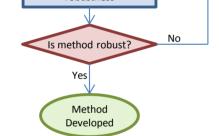
Table 1			
Column	ACE HILIC-A, ACE HILIC-B and ACE HILIC-N,		
	150 x 4.6 mm		
Isocratic screening	10 mM ammonium formate in MeCN/H ₂ O (90:10 v/v)		
	Ammonium forma	te at pH 3.0, 4.7	or 6.0.
Gradient screening	Line A: 10 mM ammonium formate in MeCN/H ₂ O (94:6 v/v)		
	Line B: 10 mM am	monium formate	e in MeCN/H ₂ O (50:50 v/v)
	Ammonium formate at pH 3.0, 4.7 or 6.0.		
	Gradient:		
		Time (mins.)	%B
		0	0
		15	100
		20	100
		21	0
		41	0

6. Example 1: Caffeine and Related Substances

- The screening protocol was used to develop a HILIC method for caffeine and related substances.
- > All analytes are polar neutral with negative logP values.



> The mixture was screened on the ACE HILIC-A, HILIC-B and HILIC-N

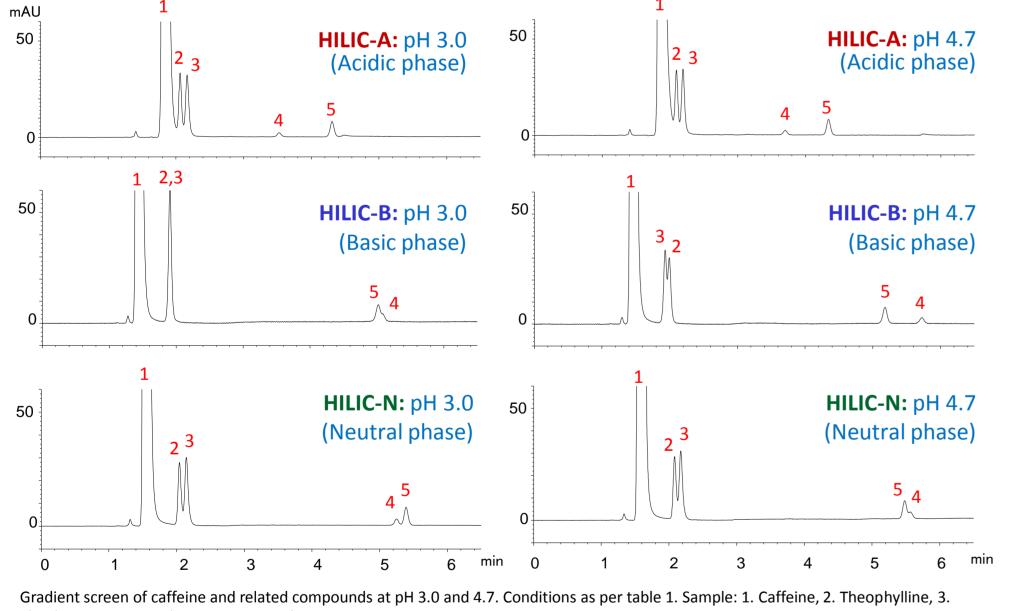


Flow rate	1.5 mL/min
Temperature	25 °C
Detection	Dependent on sample

phases at pH 3.0 and 4.7.

7. Caffeine and Related Substances: Gradient Screen

> The HILIC-N at pH 3.0 provided the most potential for a full separation.

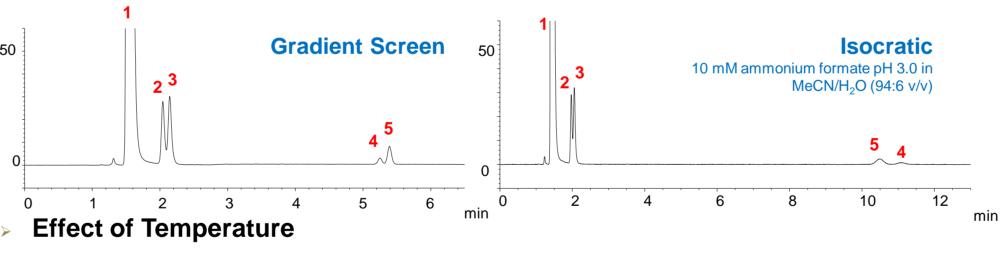


Gradient screen of caffeine and related compounds at pH 3.0 and 4.7. Conditions as per table 1. Sample: 1. Caffeine, 2. Theophylline, 3. Theobromine, 4. Xanthine, 5. Hypoxanthine First published in Chromatography Today, Volume 8, Issue 4, Nov/Dec 2015

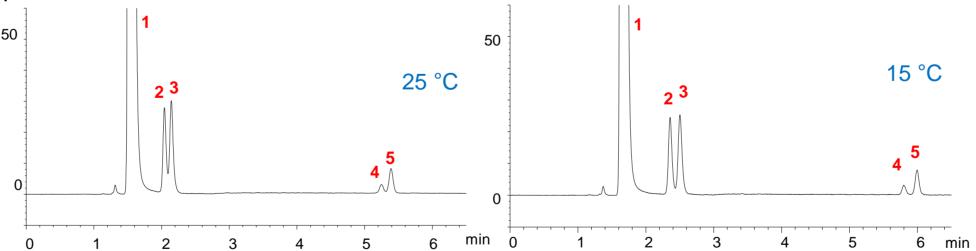
8. Caffeine and Related Substances: Optimisation

Isocratic or gradient mode

Isocratic conditions were assessed and found to increase the retention of later eluting compounds but failed to improve resolution between the critical pair.

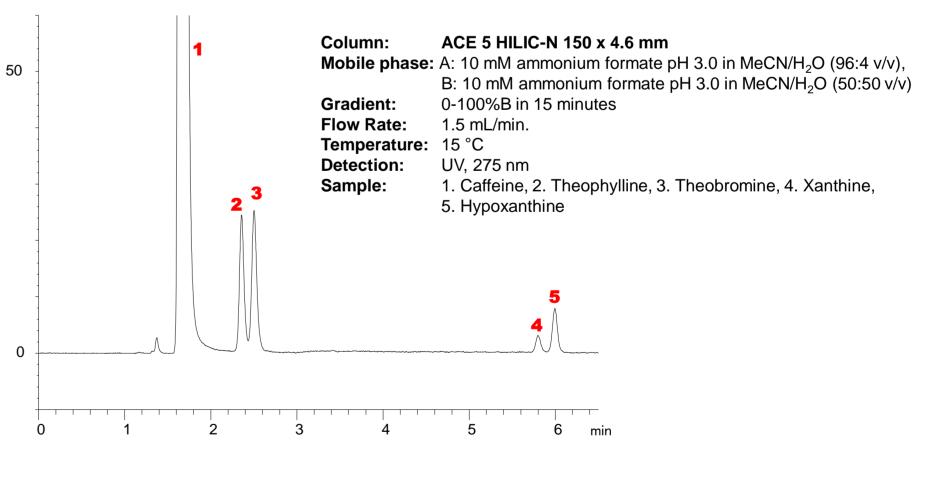


Reducing the temperature was found to improve the resolution of both impurity peak pairs.



9. Caffeine and Related Substances: Final Method

- Reduced temperature was utilised to achieve separation of the mixture on the ACE 5 HILIC-N at pH 3.0.
- A small increase in acetonitrile in the gradient starting conditions was also found to be beneficial



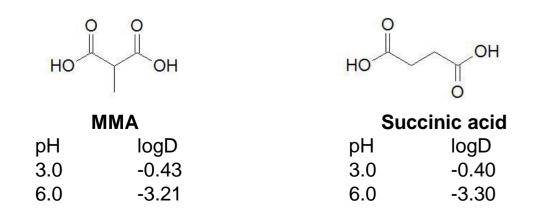
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10. Example 2: MMA and Succinic Acid

11. Example 2: MMA and Succinic Acid

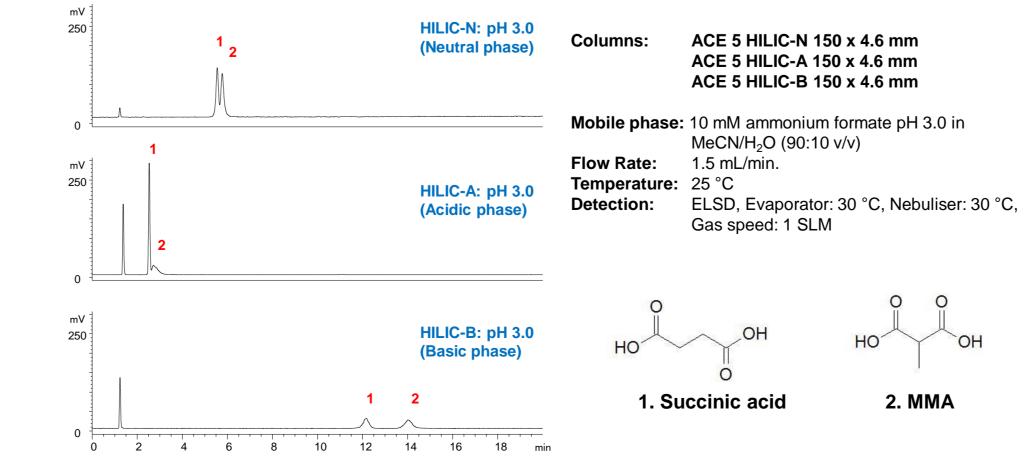
12. Summary

- Elevated levels of methylmalonic acid (MMA) in blood/urine is routinely tested for as an indicator of vitamin B12 deficiency.
- The isobaric species succinic acid may be present in biomatrices at concentrations up to 50x that of MMA.
- Chromatographic separation of these two components with good retention to eliminate matrix interference is therefore essential for accurate determination.
- > Physico-chemical properties are similar and suitable for HILIC mode



- The two compounds were screened on the three ACE HILIC phases.
- The two acids were predicted to possess a single negative charge at pH 3.0 and a double negative charge at pH 6.0, therefore these two mobile phase pH's were selected for screening

- Very strong retention was observed at pH 6.0 on all phases (data not shown).
- At pH 3.0, the analytes proved difficult to separate on the ACE HILIC-N and ACE HILIC-A phases due to similar analyte properties.
- > The ACE HILIC-B provided additional retention and selectivity for the two analytes.
- Use of the screening protocol provided a separation of these challenging analytes on the ACE HILIC-B phase with no further development required.



- A systematic and rationally designed method development strategy can aid in streamlining the method development process.
- In HILIC, column stationary phase and mobile phase pH are the most critical parameters affecting selectivity.
- The step-by-step method development strategy proposed in this poster therefore provides a powerful means by which to probe selectivity of a new application.
- The ACE HILIC-A, HILIC-B and HILIC-N phases have been shown to provide complimentary selectivity, ideal for method development.
- Screening an analyte mixture on these three phases has been demonstrated as an effective method development strategy for selecting an appropriate stationary phase/mobile phase combination.
- Optimisation can be achieved by altering parameters such as ionic strength, % organic and temperature.

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